## Second-order Nonlinearities of CdS Nanoparticles Studied by Hyper-Rayleigh Scattering Technique

## Yu ZHANG, Xin WANG, De Gang FU\*, Ju Zheng LIU, Zu Hong LU

National Laboratory of Molecular and Biomolecular Electronics, Department of Chemistry and Chemical Engineering, Southeast University, Nanjing 210096

Abstract: A series of CdS nanoparticles with different surfaces were prepared by colloidal chemical method and reverse micelle method. Their second-order nonlinear optical (NLO) properties were experimentally studied in solution by newly developed hyper-Rayleigh scattering (HRS) technique. The results show that "per particle" first-order hyperpolarizability  $\beta$  values are sensitive to the synthetic method and the surface chemical modification.

**Keywords:** Second-order nonlinear optical properties, hyper-Rayleigh scattering (HRS), first-order hyperpolarizability, CdS nanopartices, surface-modification.

Optical nonlinearities of semiconductor nanoparticles are of great interest recently. So far their third-order nonlinear optical (NLO) properties have been widely studied. However, there are only few studies on second-order NLO properties, because it is believed that the centrosymmetry or near-centrosymmetry of spherical nanoparticles eliminate their first-order hyperpolarizability  $\beta$  values to zero or near zero. And for a long time it remains a problem to directly study the second-order NLO properties of such nanoscale particles by conventional NLO technique such as interfacial second harmonic generation (SHG) and electric-field-induced SHG techniques which are constrained by the orientational, size, and/or charge restrictions<sup>1</sup>. Fortunately, the newly developed hyper-Rayleigh scattering (HRS) technique overcomes the above restrictions, hence second-order NLO properties of nanoparticles can be studied. Recently, a few studies were reported about HRS for the colloidal gold and insulator nanoparticle SiO<sub>2</sub><sup>1,2</sup>. Here the HRS technique is used to measure second-order NLO response of a series of semiconductor nanoparticles with different surfaces prepared by different methods.

The  $Cd^{2+}$ -rich CdS,  $S^{2-}$ -rich CdS, thiourea-stabilized  $Cd^{2+}$ -rich CdS and heteropolyanion-stabilized  $Cd^{2+}$ -rich CdS hydrosols are named as  $CdS/Cd^{2+}$ ,  $CdS/S^{2-}$ ,  $CdS/SC(NH_2)_2$  and  $CdS/PW_{12}O_{40}^{3-}$ , prepared by rapidly mixing  $Cd(NO_3)_2$  and  $Na_2S$ aqueous solutions under stirring. Using reverse micelle method<sup>3</sup>, the surface-modified CdS nanoparticles with  $Cd^{2+}$ -rich surface by AOT (bis (2-ethylhexyl) sulfosuccinate, disodium salt) or pyridine (Py) were synthesized (named as CdS/AOT and CdS/Py, respectively). The absorption spectra and TEM images show that the above CdS nanoparticles are about 5 nm in diameter with narrow size distribution and have Yu ZHANG et al.

negligible absorption at the frequency-doubling light of 532 nm.

The HRS experiments use a similar setup to the literature<sup>4</sup>. The Q-switched Nd-YAG laser pulse (10 Hz and 8-10 ns pulse width) at 1064 nm is focused into a 5 cm length glass cell in which a liquid sample is measured. The  $\beta$  values of nanoparticles are determined in terms of the internal reference method (IRM)<sup>1</sup>. The calculation shows that the  $\beta$  values of the CdS/Cd<sup>2+</sup>, CdS/S<sup>2-</sup>, CdS/SC(NH<sub>2</sub>)<sub>2</sub>, CdS/PW<sub>12</sub>O<sub>40</sub><sup>3-</sup> in water and CdS/AOT in heptane, CdS/Py in pyridine are  $1.30 \times 10^{-26}$ ,  $1.41 \times 10^{-26}$ ,  $4.38 \times 10^{-26}$ ,  $2.51 \times 10^{-26}$  and  $1.50 \times 10^{-27}$ ,  $1.05 \times 10^{-27}$  esu, respectively, which are among the largest values reported for solution species. It is found that the  $\beta$  values differ by one order of magnitude for the CdS nanoparticles prepared with the different methods. And the CdS nanoparticles with different surfaces also have different  $\beta$  values. As proved by Clays *et al.*<sup>5</sup>, surface termination of the crystalline lattice creates a condition of noncentrosymmetry which contributes to the large  $\beta$  values for nanoparticles with huge surface-to-volume ratio. It is apparent that synthetic methods and surface-modifications have strong influence on the surface structure of CdS nanoparticles. Therefore,  $\beta$  values of CdS nanoparticles can be controlled. This is significant for further detailed investigation of HRS mechanism of nanoparticles.

## Acknowledgments

This work is supported by The National Natural Science Foundation of China (No.59582005).

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Received 15 November 1999 Revised 26 January 2000